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Question 13 The oxidation of sulfur dioxide is an exothermic reaction, as shown in the equation below. The reaction is catalysed by vanadium(V) oxide. $2\text{SO}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{SO}_3(\text{g})$

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Question 1 A catalyst A. slows the rate of
reaction. B. ensures that a reaction is
exothermic. moves the chemical

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equilibrium of a reaction in the forward direction.

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